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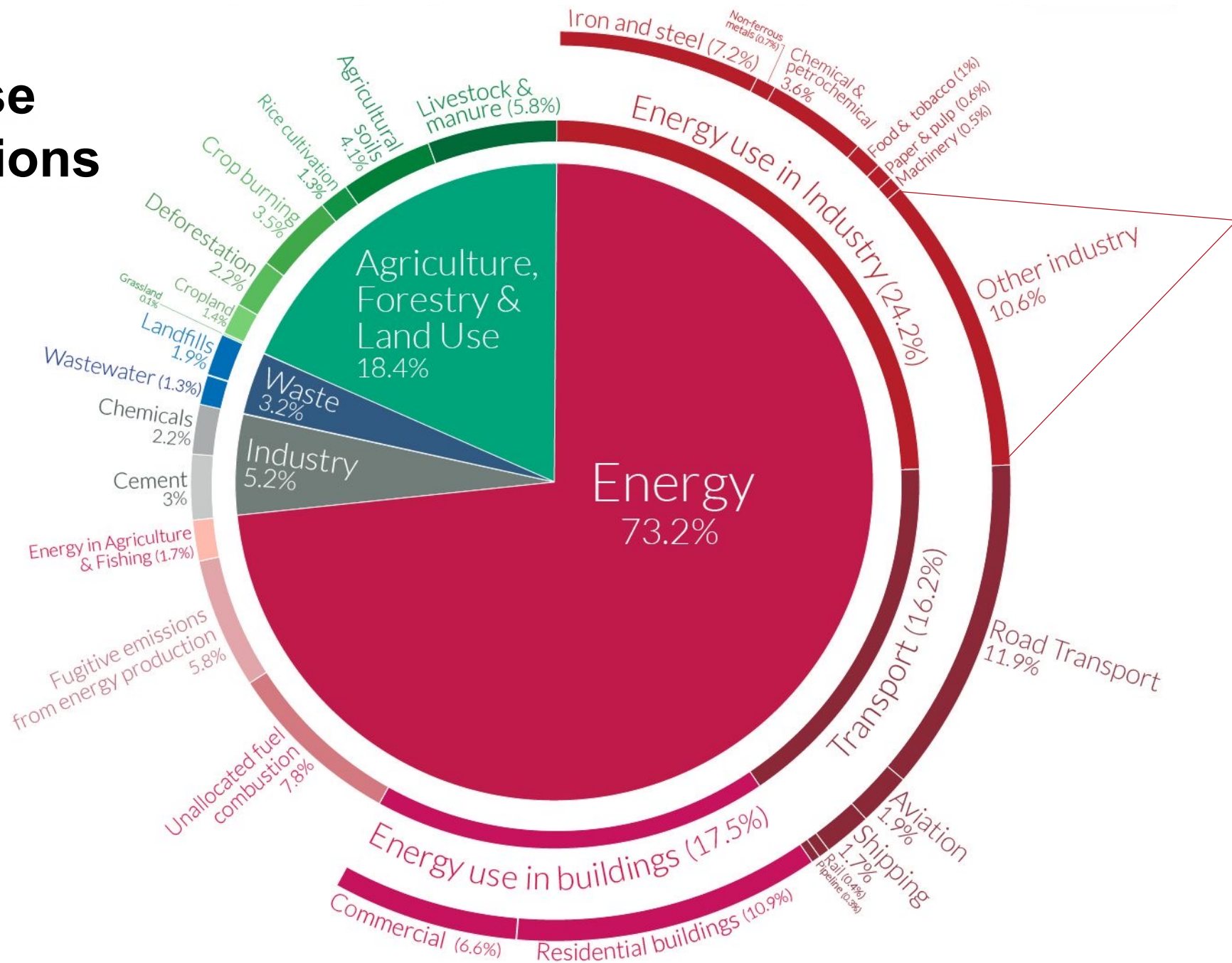
Potential for atmospheric CO₂ removal in mafic quarries via enhanced weathering of basalt fines

Amanda Stubbs (a.stubbs.1@research.gla.ac.uk), Faisal Khudhur, John MacDonald, Linzi McDade, and Mark Friel



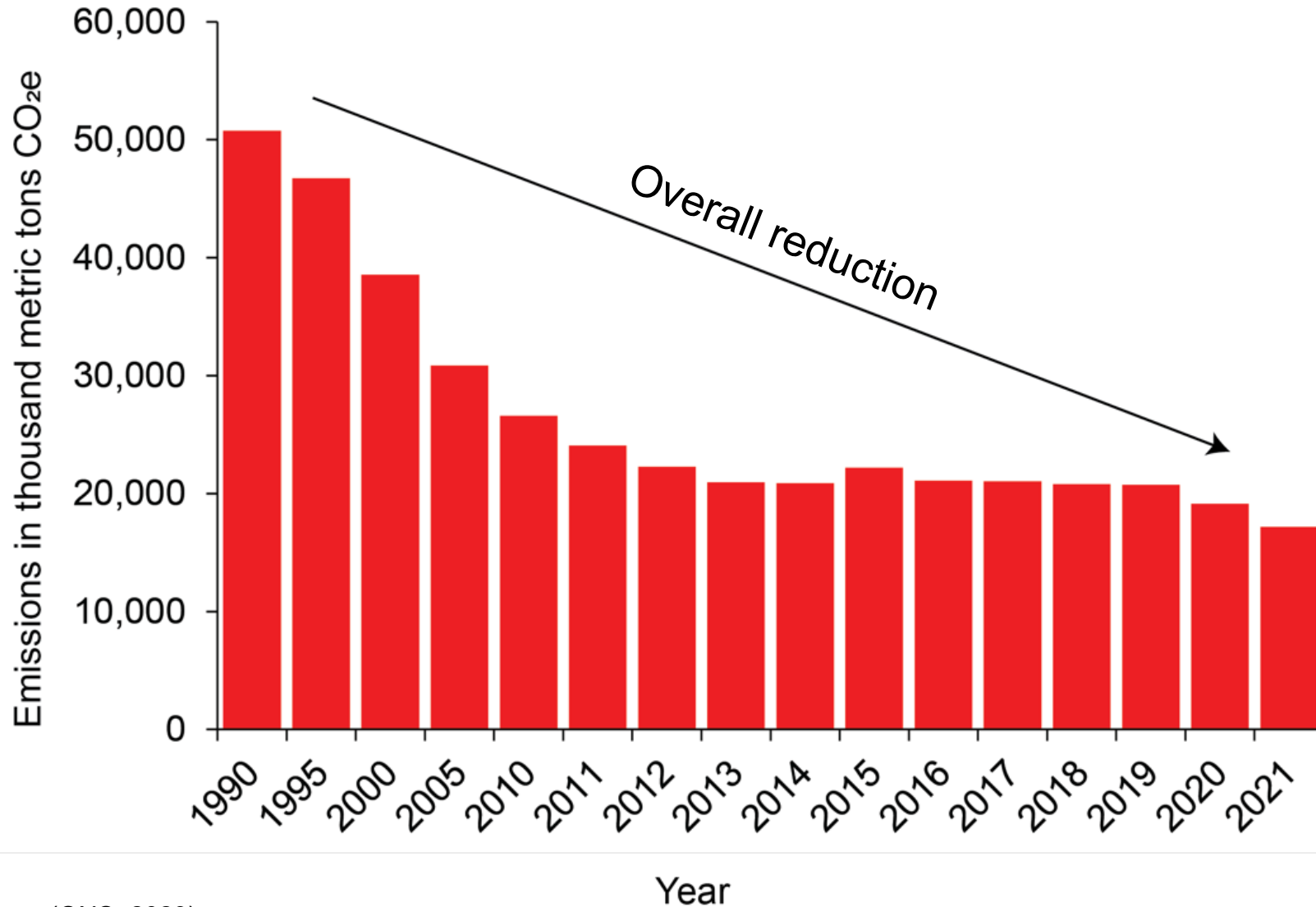
Global greenhouse gas emissions by sector

Climate Watch, the World Resource Institute. (2020)



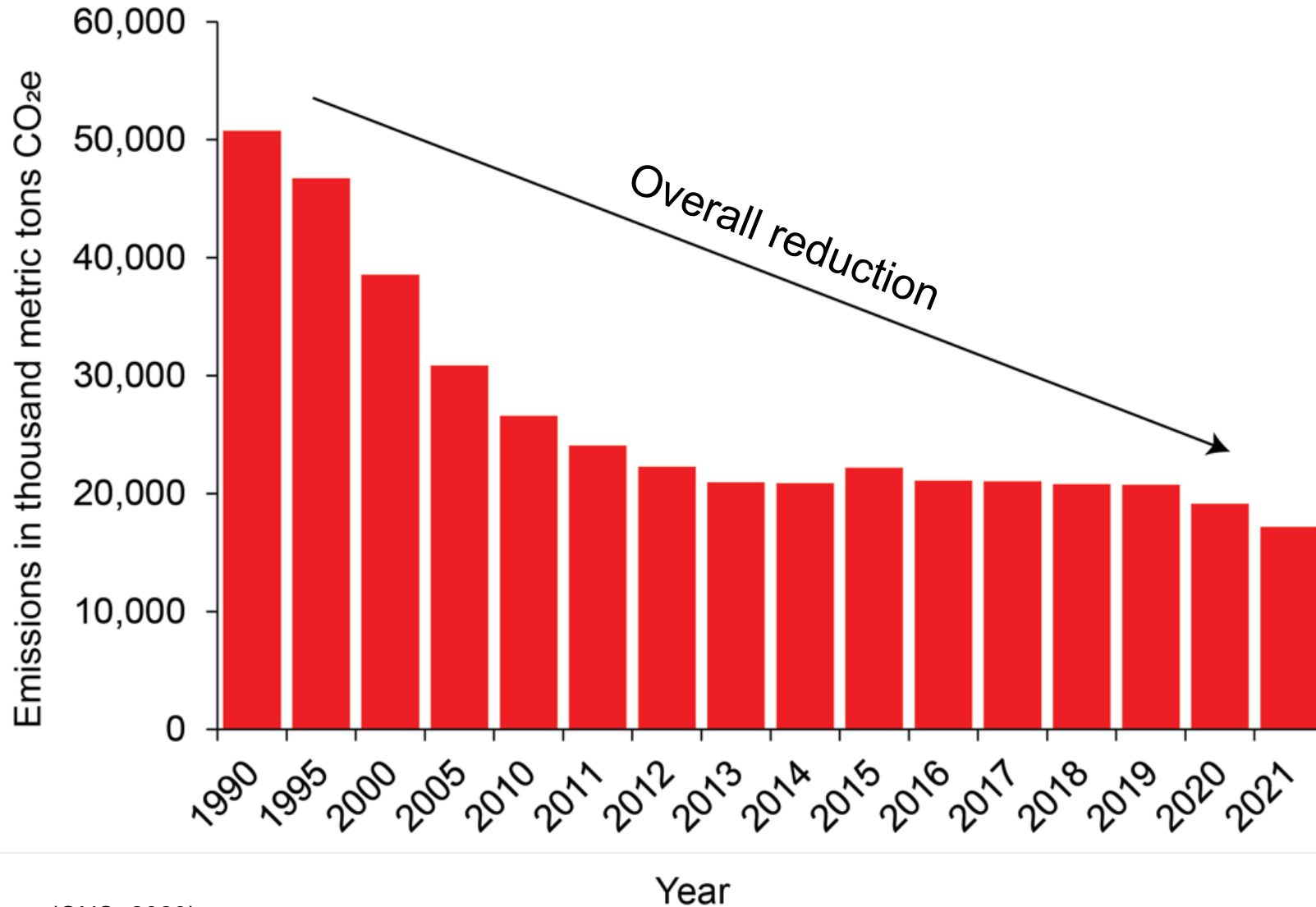
Mining & quarrying operations: 4 to 7%

UK greenhouse gas emissions in the mining and quarrying industry



- Since the turn of the century, over a 40% reduction
- CO₂ accounted for more than 90% of greenhouse gas emissions (ONS, 2023)

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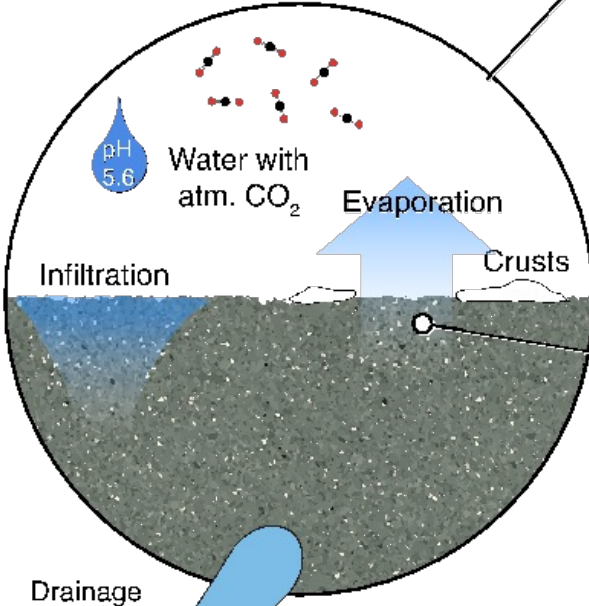
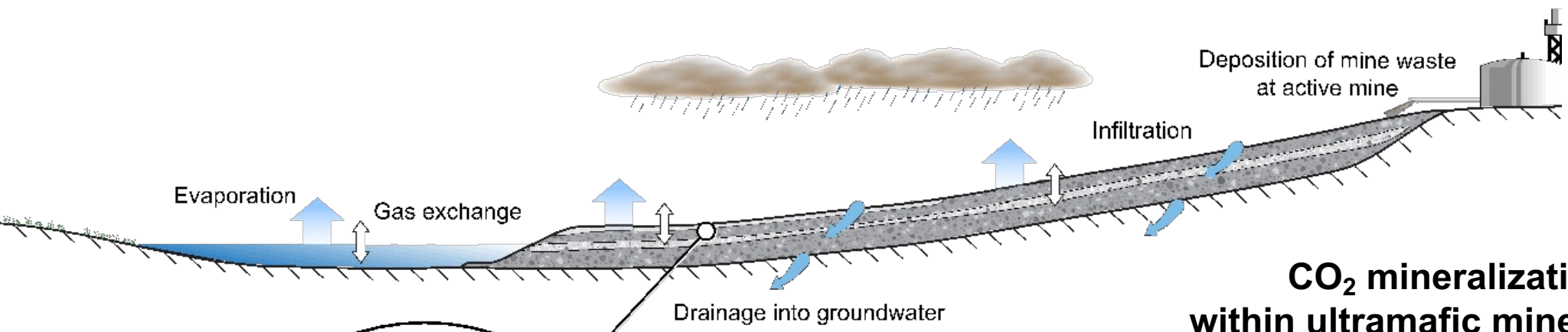


(ONS, 2023)

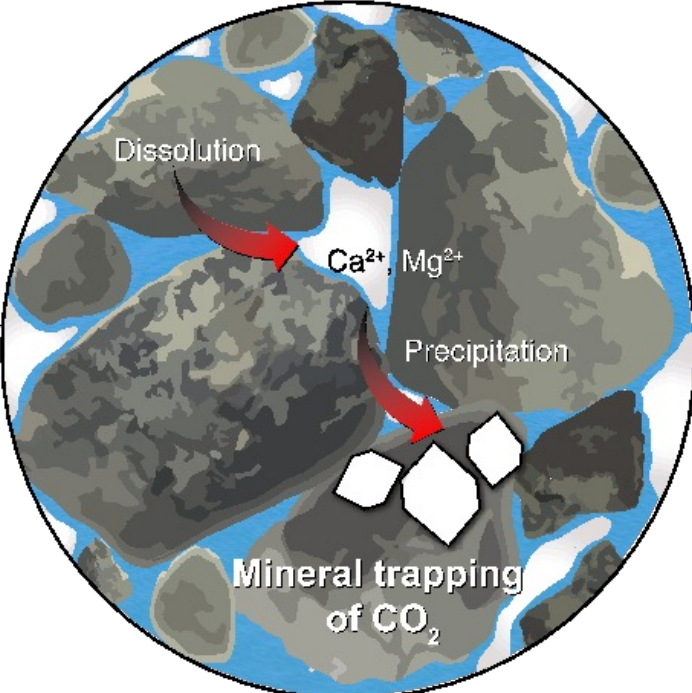
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- More sustainable practices**
- Modifying equipment design
 - Optimising resource use
 - Using renewable resources

Negative emission technologies: Enhanced weathering of quarry fines



Solubility trapping of CO₂



CO₂ mineralization within ultramafic mine wastes:

- Wilson et al., 2009
 - Assima et al., 2012
 - Beudoin et al., 2017
 - Turvey et al., 2018
- } Chrysotile
- Wilson et al., 2011
 - Stubbs et al., 2022
 - Stubbs et al., 2023
- } Kimberlite

Advantages and disadvantages of utilising quarry fines

Strengths

- In 2018, the UK produced 80.4 million tonnes of mineral waste (Department of environment, food, and rural affairs, 2023)
- Quarries have the capability to move waste around sites
- Primarily fine-grained material
- Desirable geochemical composition
- Financial backing

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Opportunities

- 1750 active quarries in the UK, many extract mafic rock (Cameron et al., 2020)
- In Scotland alone there are 37 mafic quarries
 - Basalt, dolerite, and/ or gabbro (Cameron et al., 2020)
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- Currently no management practices in place
- Limited research/ knowledge about on-site removal

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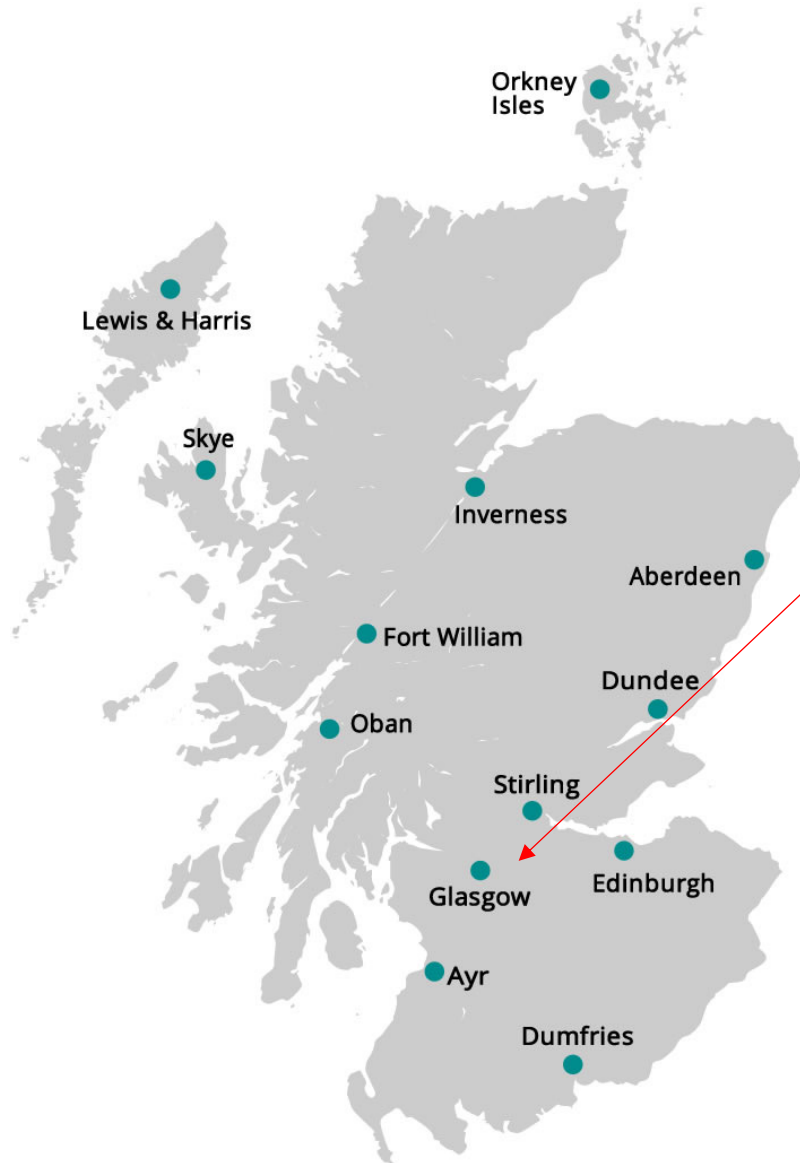
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Threats

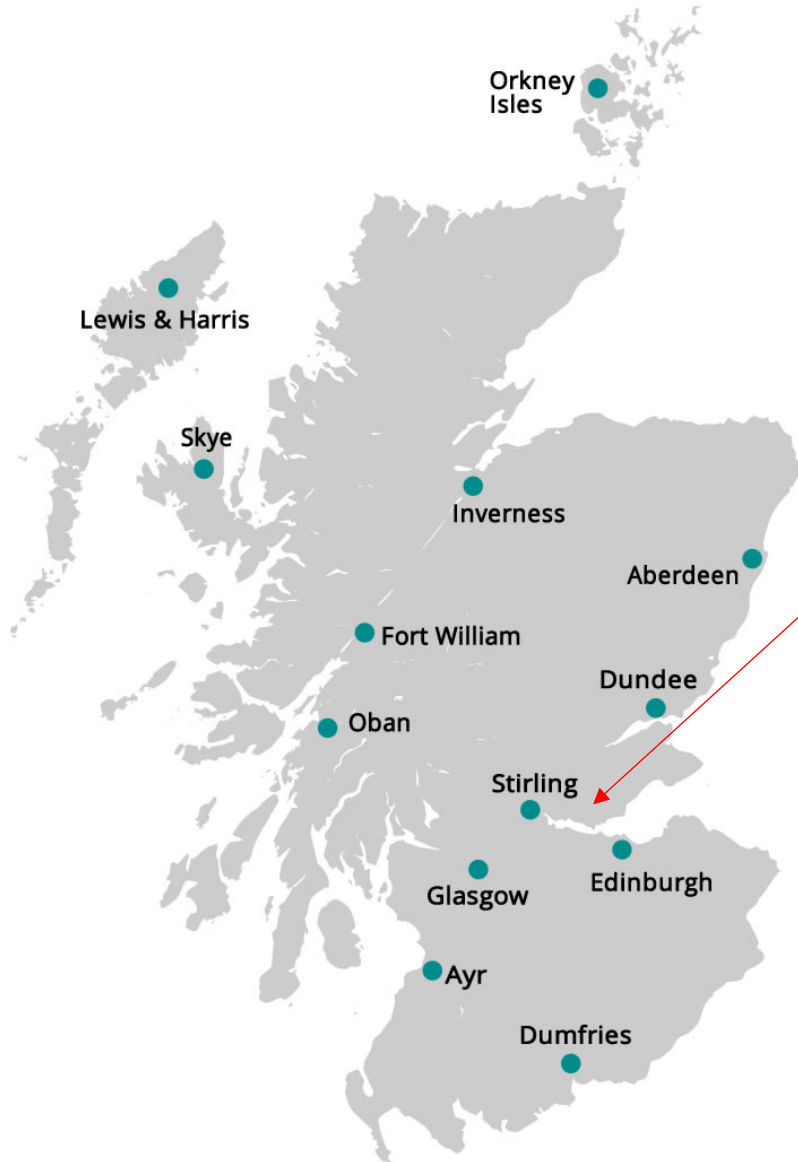
- Release of ecotoxic elements
- More time and resources researching traditional mine waste as a feedstock
 - Tailings from active and historical:
 - Diamond mines
 - Nickel mines
 - Asbestos mines

Fieldwork and sampling: Cairneyhill Quarry



- Established in the 1950s
- Asphalt production for road building and construction
- Exploiting large microgabbro/ dolerite
- More than 83,100 tonnes of available fines

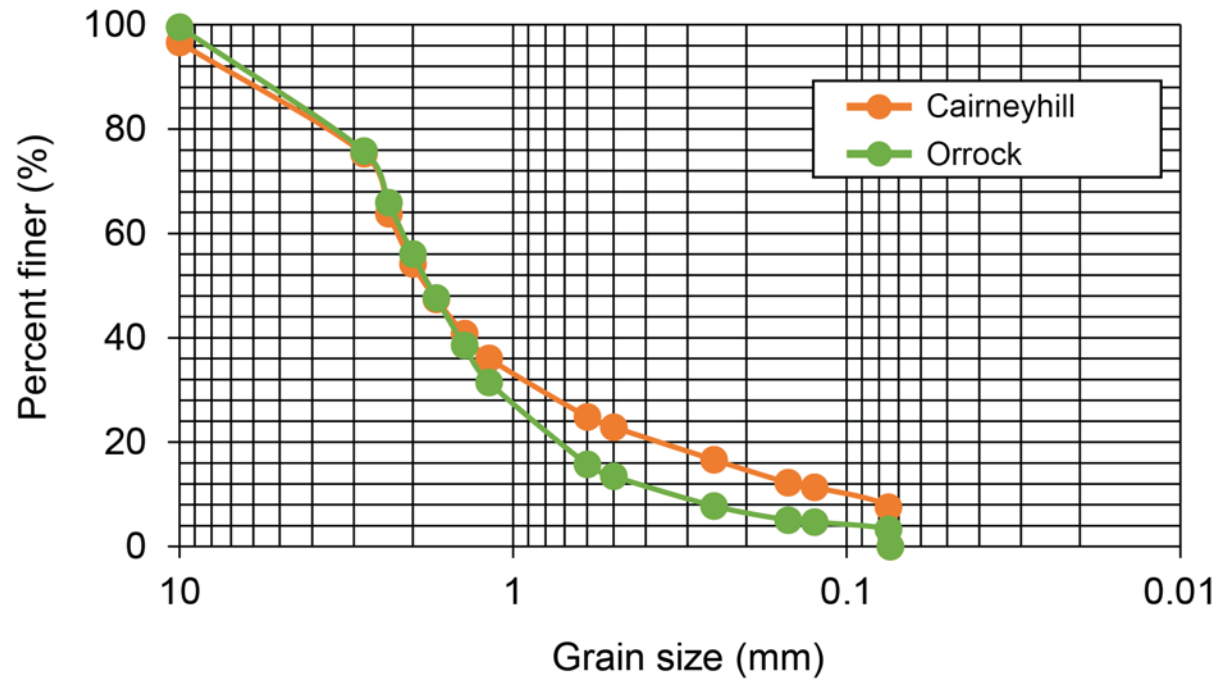
Fieldwork and sampling: Orrock Quarry



- Established in the 1950s
- Asphalt and aggregate production
- Exploiting basalt
- 42,000 to 56,000 tonnes of fines annually



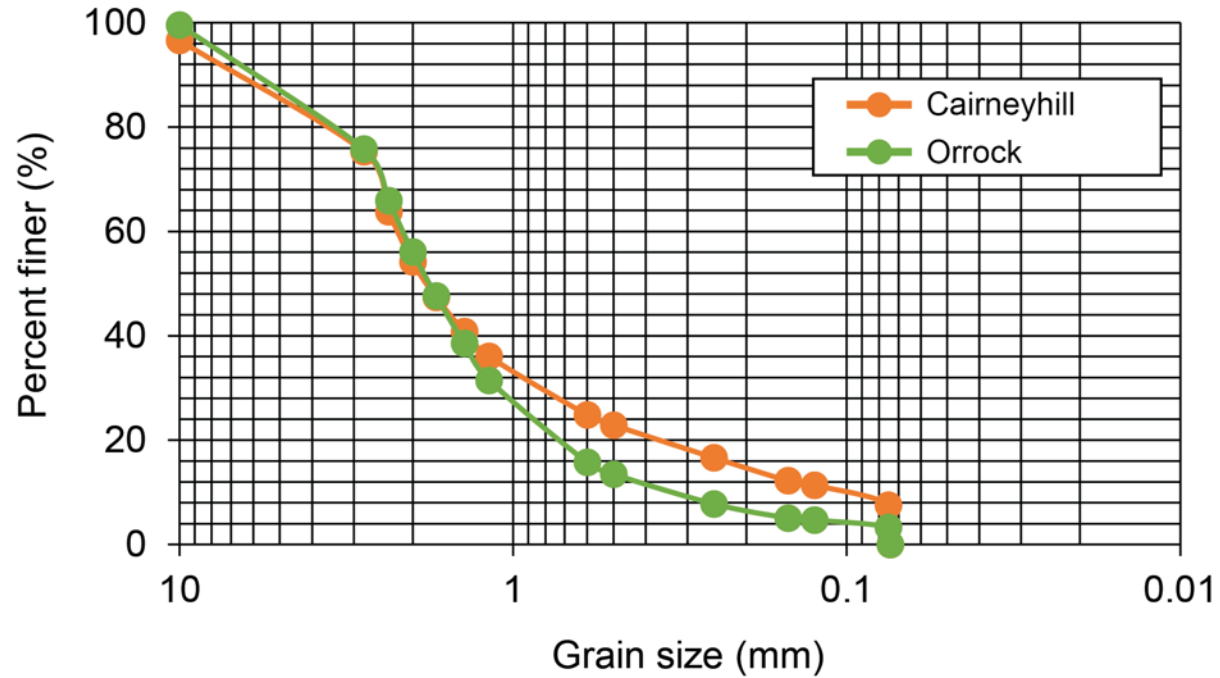
Initial characterisation: Physical properties



Both fines are dominated by sand sized particles



Initial characterisation: Physical properties



Both fines are dominated by sand sized particles

BET surface area

Cairneyhill = 3.05 m²/g

Orrock = 5.82 m²/g

Initial characterisation: Mineralogical properties

Cairneyhill Quarry

Target Minerals	Abundance (wt.%)	Desired cation
Anorthite	50.0	Ca
Augite	19.8	Ca, Mg
Chlorite	10.7	Mg
Lizardite	4.2	Mg

Orrock Quarry

Target Minerals	Abundance (wt.%)	Desired cation
Augite	34.9	Ca, Mg
Andesine	20.0	Mg
Forsterite	8.8	Mg
Chlorite	5.5	Mg

Initial characterisation: Geochemical properties (XRF)

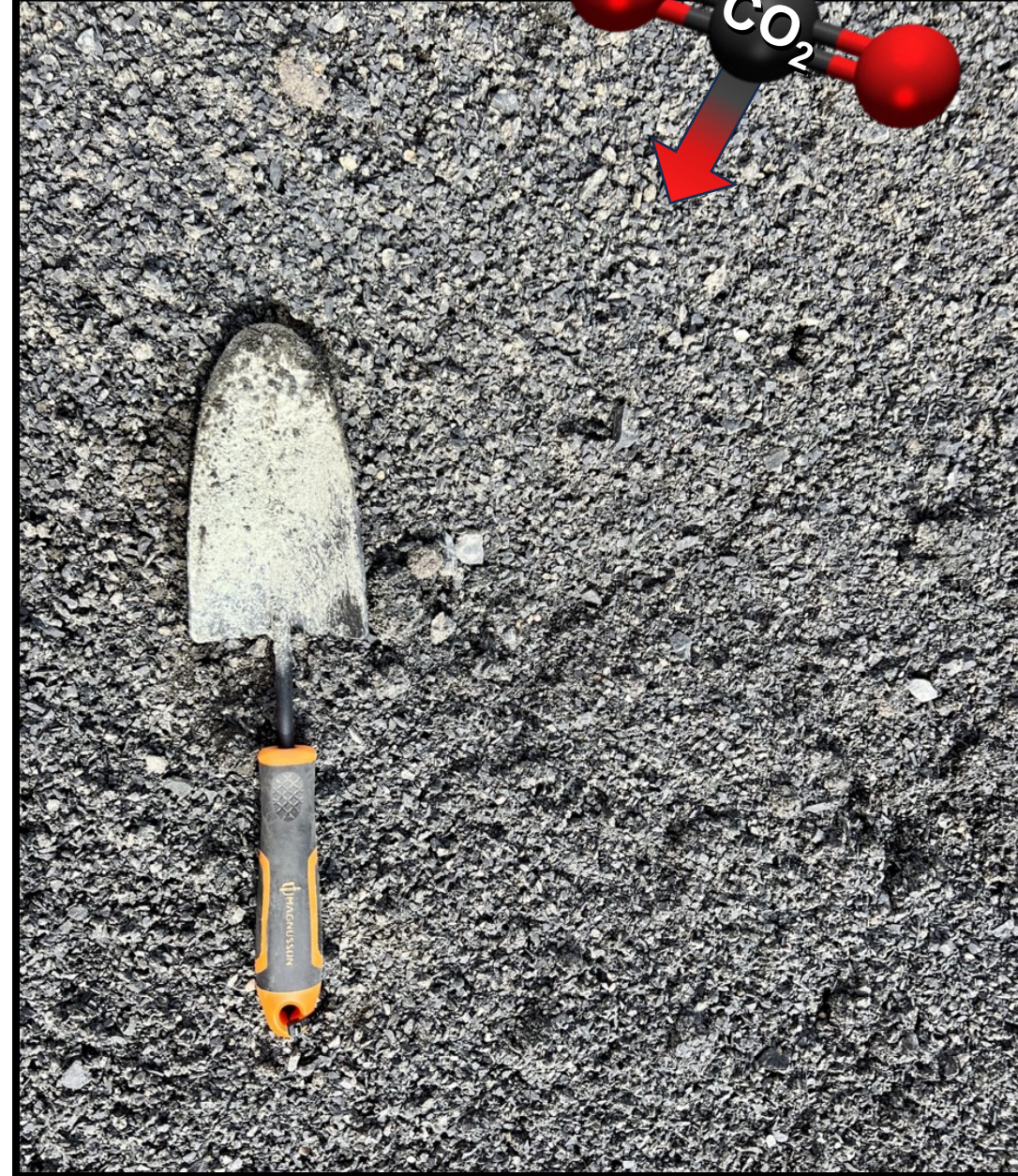
Cairneyhill Quarry:

- CaO = 8.5%
- MgO = 8.7%
- P₂O₅ = 0.2%

Orrock Quarry:

- CaO = 9.7%
- MgO = 10.3%
- P₂O₅ = 0.6%

Carbonation potential calculated based on Steinour formula from Gunning et al., 2010 and Renforth 2019



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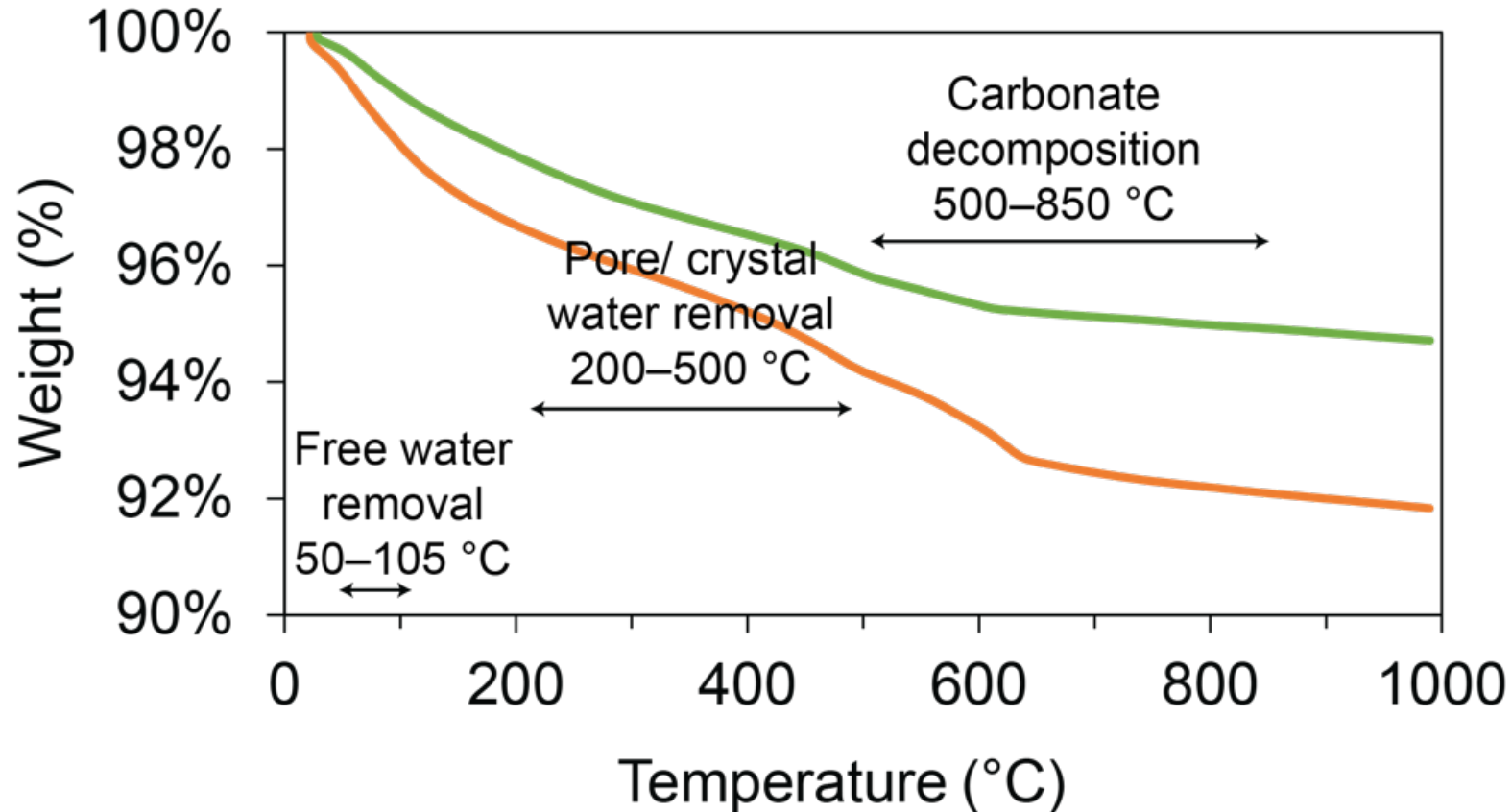
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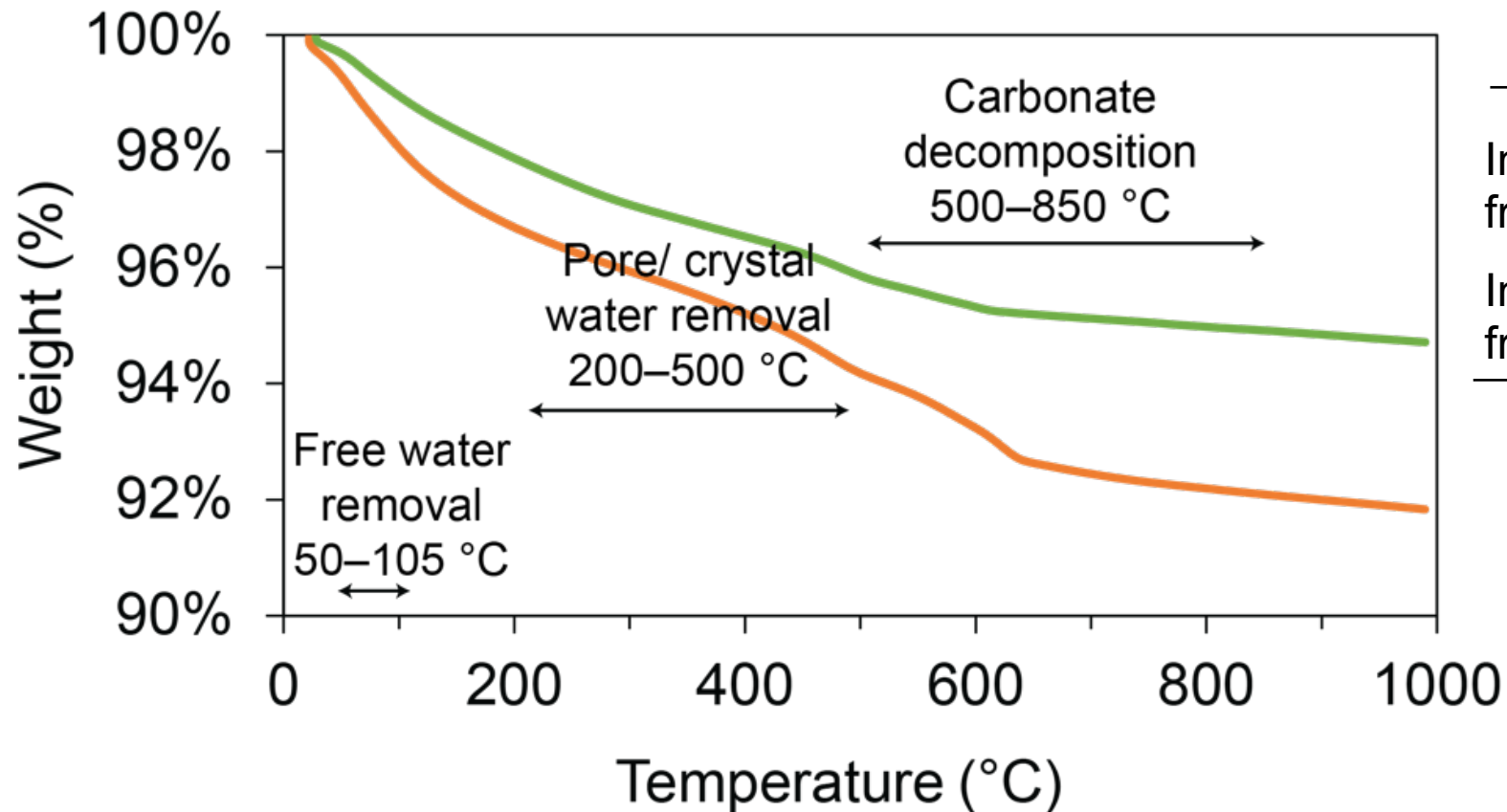
Thermogravimetric analysis



- Cairneyhill Quarry
- Orrock Quarry

Initial characterisation: Geochemical properties

Thermogravimetric analysis



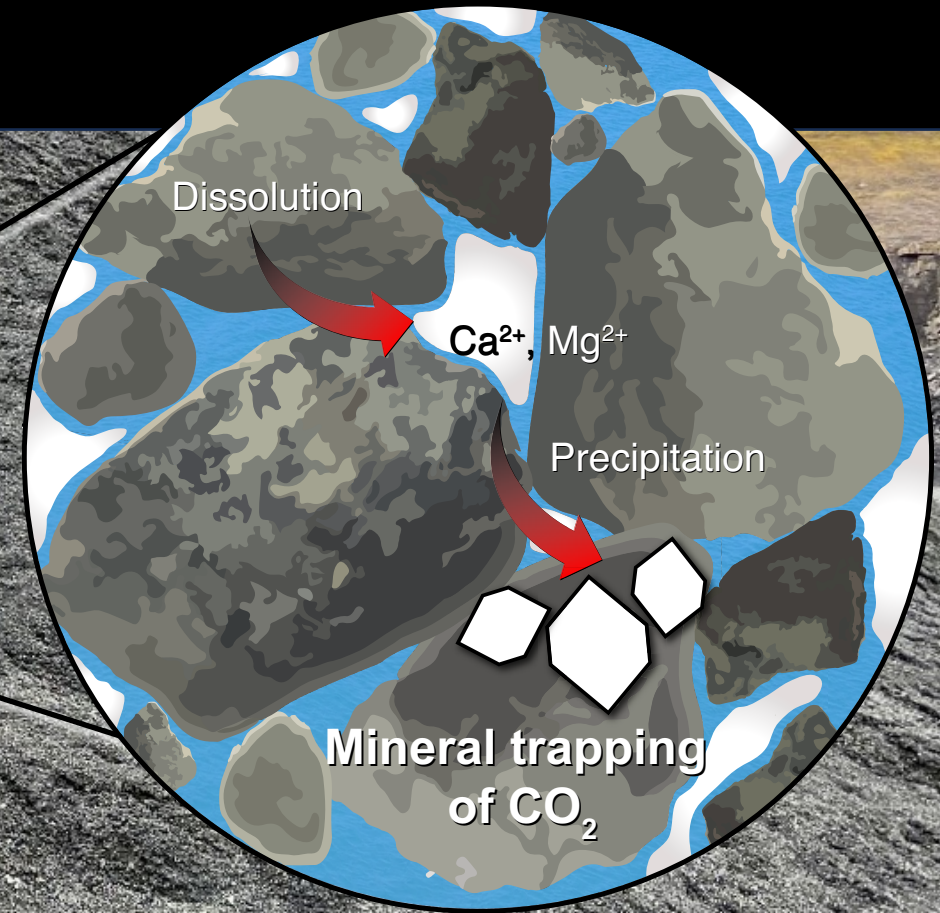
	Cairneyhill	Orrock
Initial CO ₂ (%) from TGA	2.22	1.02
Initial CO ₂ (%) from TIC	1.17	0.22

- Cairneyhill Quarry
- Orrock Quarry

Estimation of passive CO₂ sequestration

23 kg CO₂/ t Cairneyhill fines

10 kg CO₂/ t Orrock fines



An aerial photograph of a large-scale quarry operation. The landscape is dominated by dark, layered rock formations and numerous piles of processed material in various shades of grey and brown. A network of dirt roads winds through the site. In the lower right, several pieces of heavy machinery, including a red conveyor system, a green truck, and a white truck, are visible. The sky is filled with large, white, fluffy clouds, suggesting a bright but slightly overcast day.

How can the management of quarry fines be optimised for on-site CO₂ removal?

Basalt carbonation experiments

- **Comparative analysis to assess limiting carbonation factors**
- Two carbonation conditions:
 1. Ambient UK conditions
 2. Engineered approach to accelerate weathering
- Two different grain sizes:
 1. Bulk fines (no modifications made)
 2. $<75\ \mu\text{m}$
- Two different depths:
 1. 1 cm
 2. 5 cm



Basalt carbonation experiments

- Comparative analysis to assess limiting carbonation factors
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Basalt carbonation experiments

MMM Medcenter Climacell EVO Climate Chamber

- 10 °C (average annual UK temperature)
- 0.04% CO₂ (ambient conditions)
- 80% relative humidity (average annual UK RH)

Thermo Fisher Scientific Series 8000 Direct-Heat CO₂ Incubator

- 50 °C (very high end of ambient conditions)
- 20% CO₂ (engineered approaches)
- 80% relative humidity (average annual UK RH)



Basalt carbonation experiments

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- **Two different grain sizes:**
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 2. **<75 μm**
- Two different depths:
 1. 1 cm
 2. 5 cm



Basalt carbonation experiments

- Bulk fines (no human modifications made)
 - **Benefits:** Less time, energy and cost on further processing
 - **Drawbacks:** Generally, has a lower reactive surface area

- Sieved fines (<75 μm)
 - **Benefits:** Increased reactive surface area
 - **Drawbacks:** Further processing is required, therefore greater emissions output



BET surface
area

3.06 m^2/g



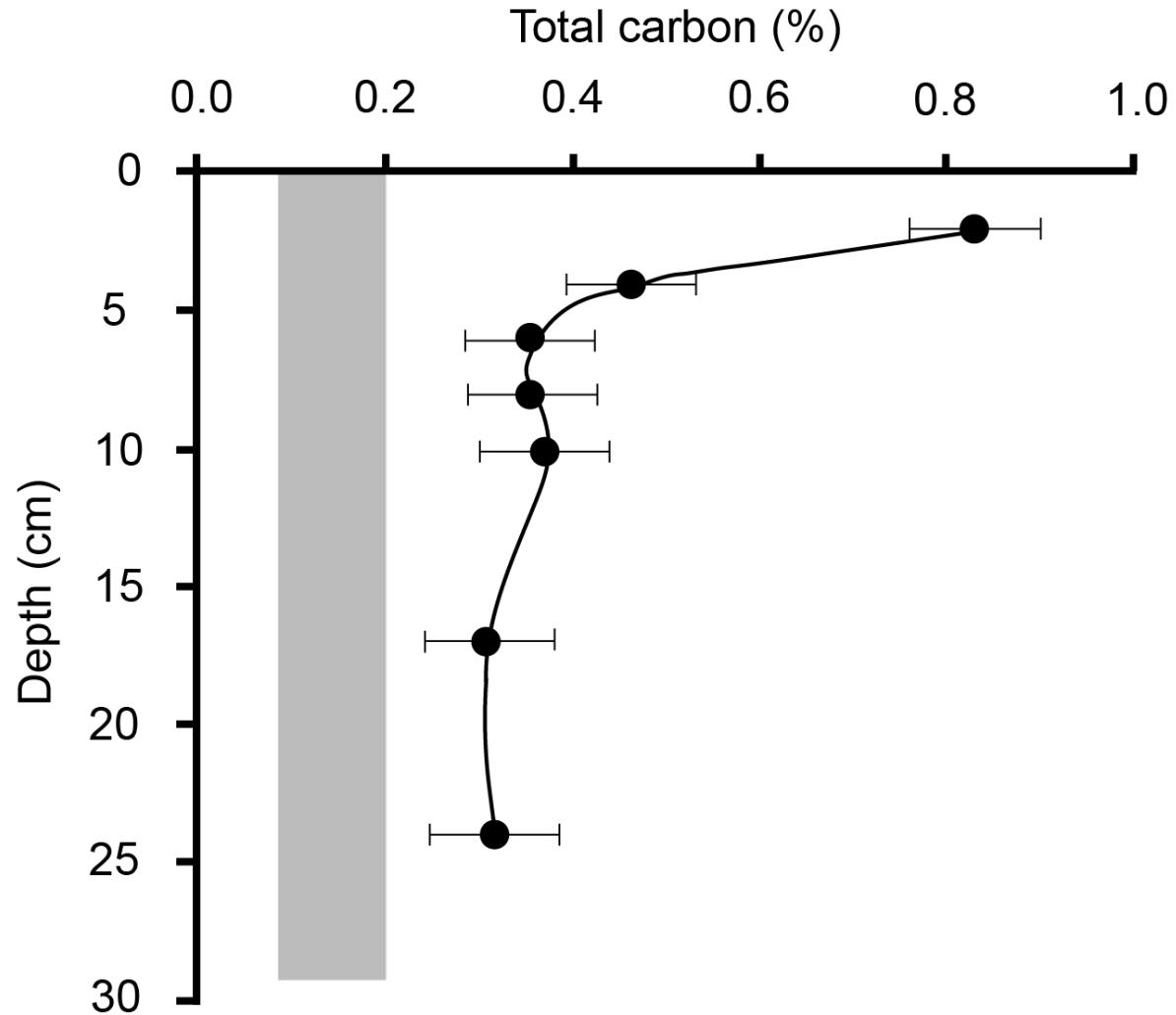
18.39 m^2/g

Basalt carbonation experiments

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- Two different grain sizes:
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- **Two different depths:**
 1. 1 cm
 2. 5 cm



Basalt carbonation experiments



(Kandji et al., 2017)

1 cm depth:

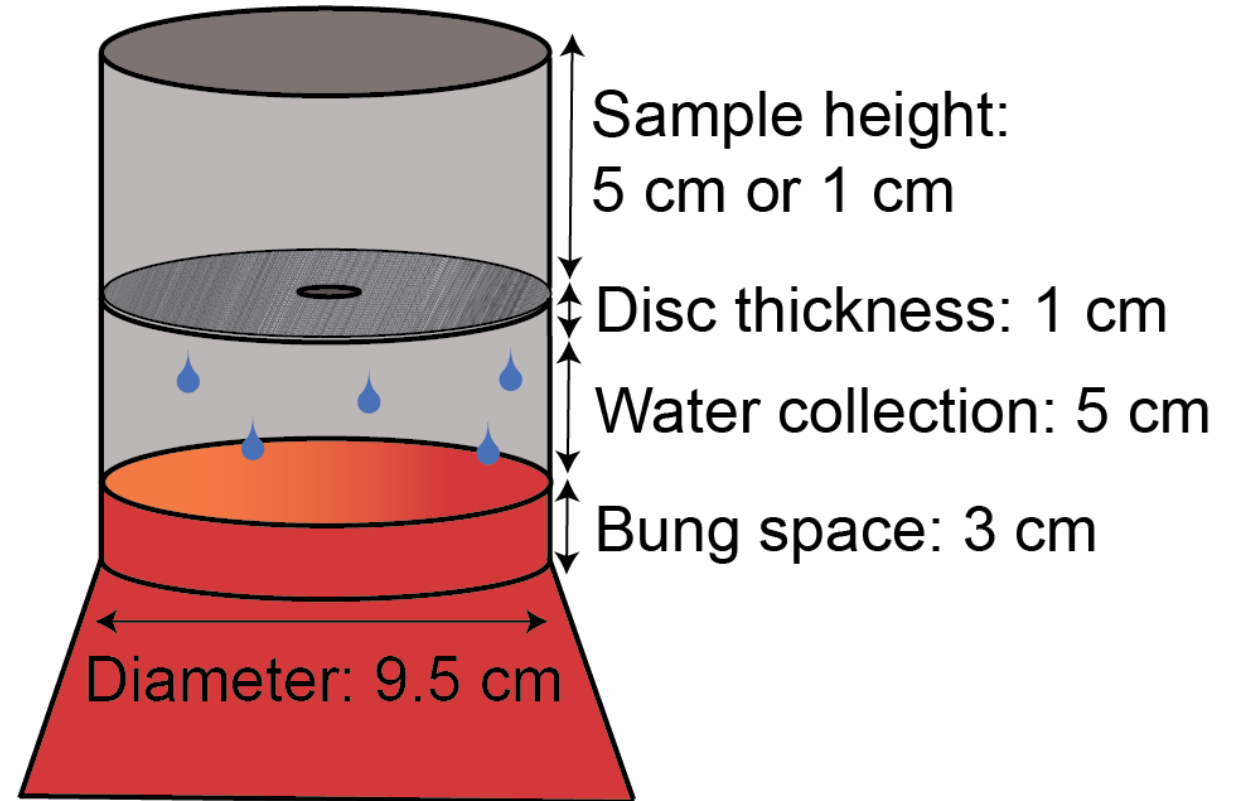
- Greater carbonation in the top surface of tailings
- CO₂ diffusion may not be a limiting factor

5 cm depth:

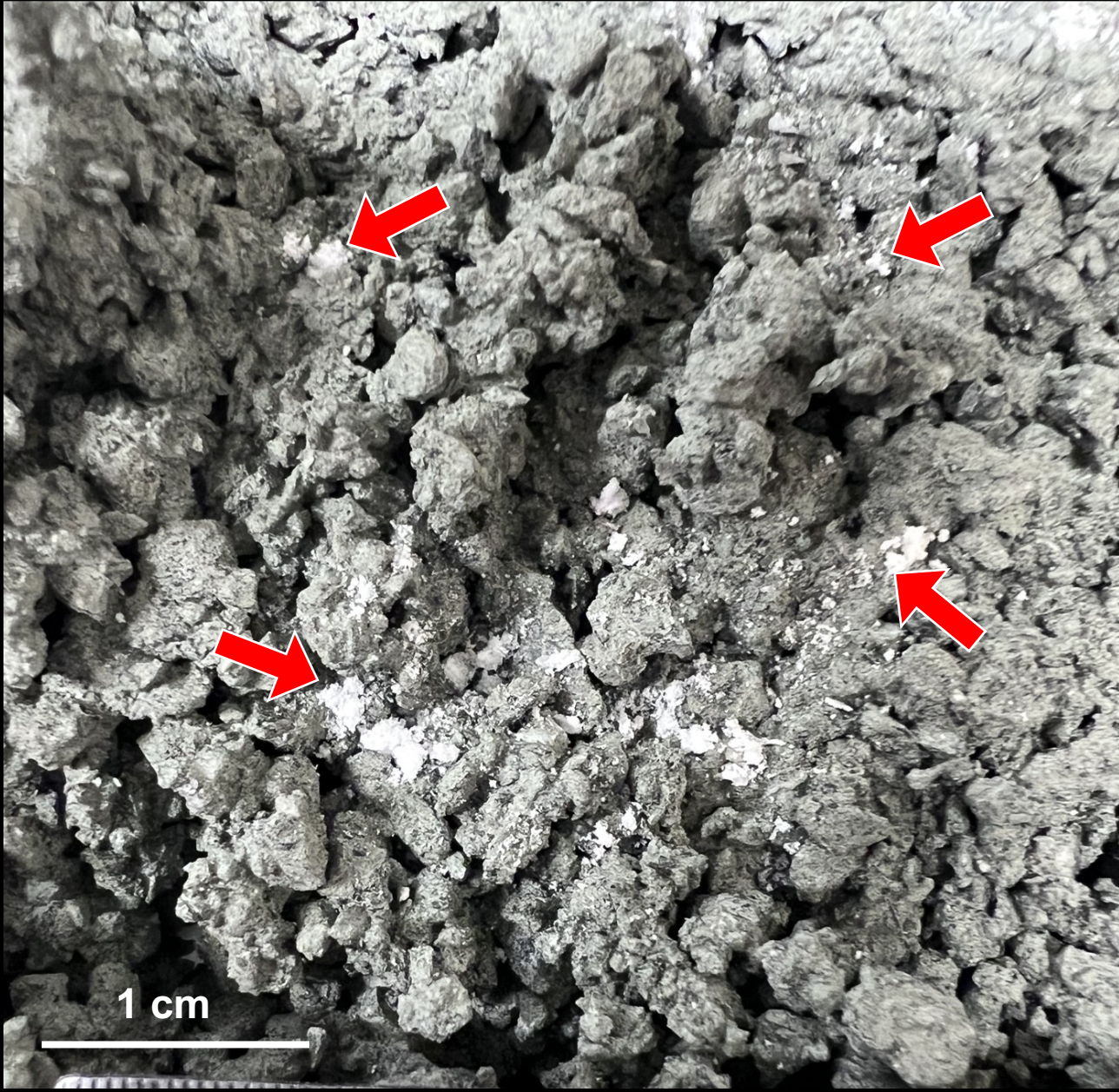
- Generally, carbonation rates decrease after this depth

Basalt carbonation experiments

- Column experiments
 - Bulk fines:
 - 1 cm = 100 g
 - 5 cm = 500 g
 - <75 μm fines:
 - 1 cm = 70 g
 - 5 cm = 500 g
- Monthly water addition equivalent to 60% pore water saturation
- Disc with 25 μm nylon mesh in the middle of the column to separate fines from water
- 3 months total

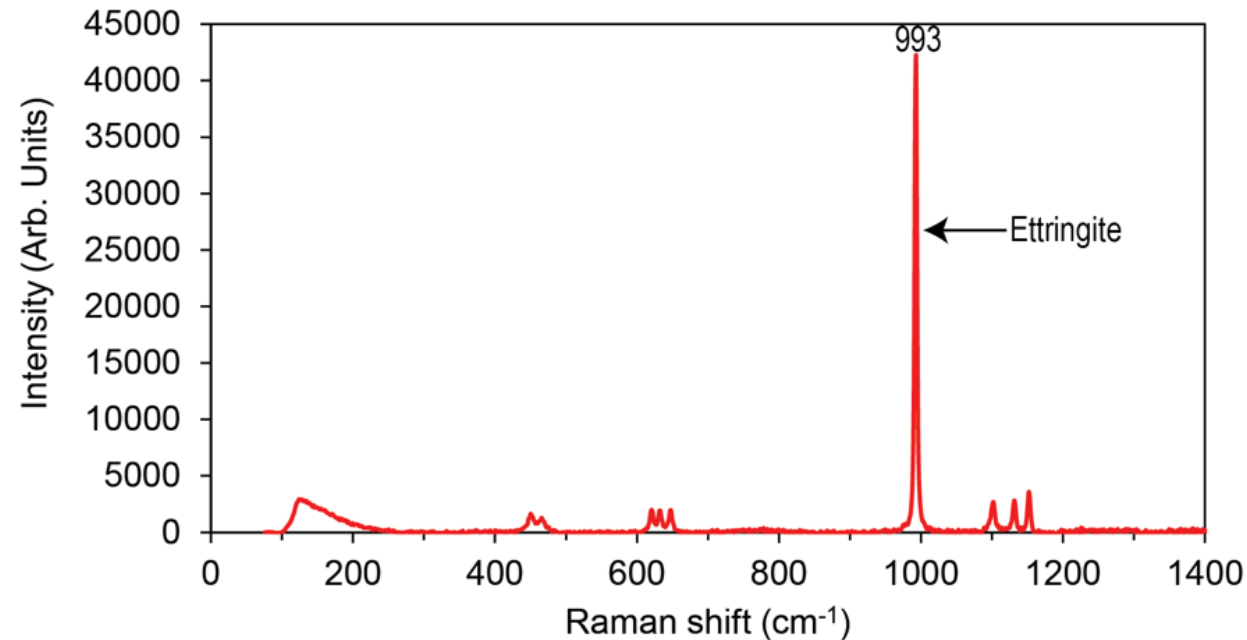


Preliminary results



White efflorescence's observed on the surface during initial drying

- Fizzed with dilute HCl
- Raman spectroscopy suggests these are Ca-bearing sulphates (i.e., Ettringite)

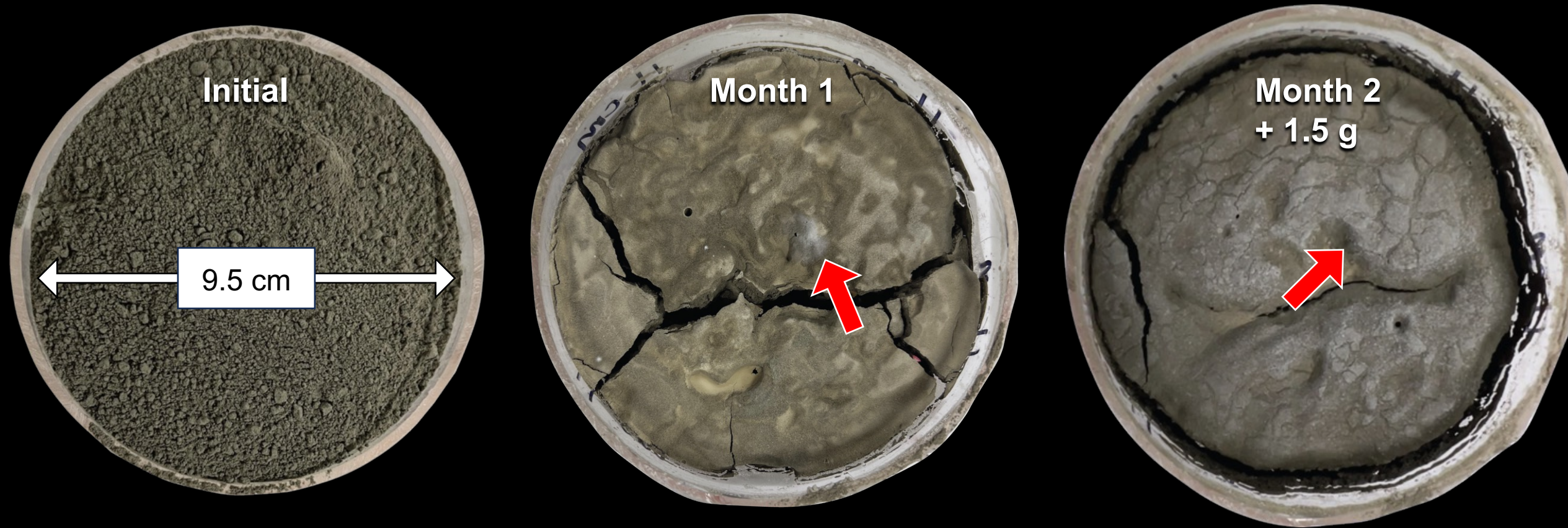


Preliminary results: Bulk fines – Ambient UK conditions



Orrock Quarry bulk fines (5 cm depth) – 4 kg CO₂/ t Orrock fines/ yr

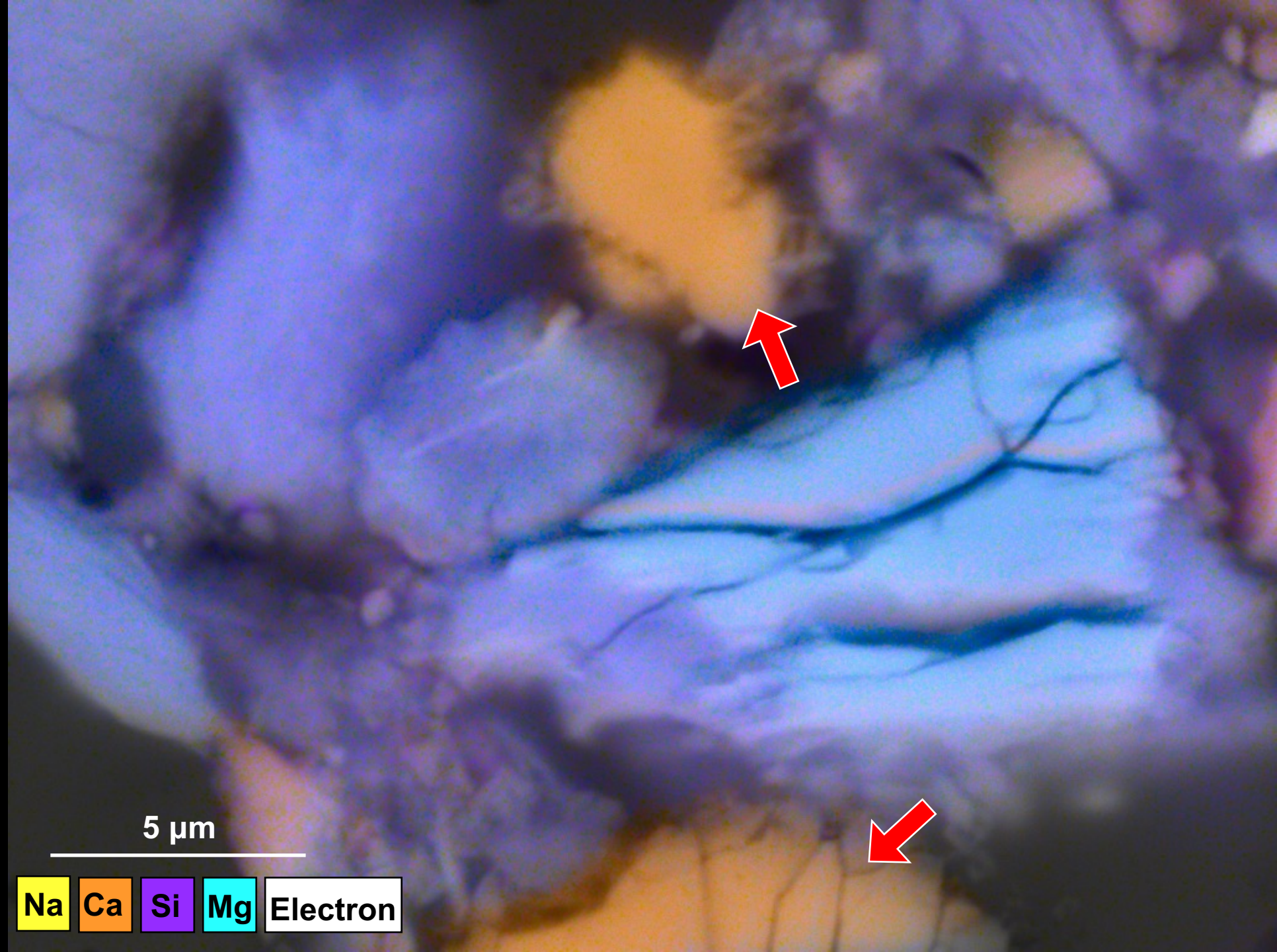
Preliminary results: <75 μm fines – Engineered approach



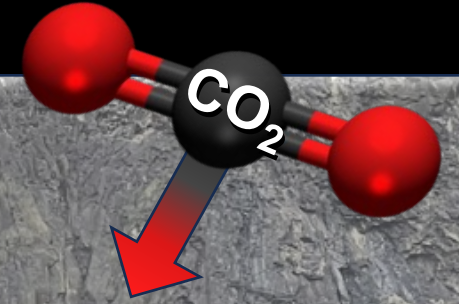
Orrock Quarry <75 μm fines (5 cm depth) – 17 kg CO₂/ t Orrock fines/ yr

**Preliminary
results: <75 μm
fines –
Engineered
approach**

 **Calcium
carbonates**



Implications for CO₂ sequestration



- **Potential for quarries to offset their CO₂ emissions**
- **Engineered approach (<75 μm fines):**
 - 840 g CO₂/ m²/ yr
- **Ambient UK conditions (Bulk fines):**
 - 276 g CO₂/ m²/ yr
 - Considering total fines production (42,000 t) and operation time remaining at Orrock (6 yrs):
 - 1006 t CO₂/ yr

Acknowledgements

Email: a.stubbs.1@research.gla.ac.uk

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